

4-[(2-Hydroxybenzylidene)amino]-5-(2-thienylmethyl)-2H-1,2,4-triazol-3(4H)-one

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Key indicators

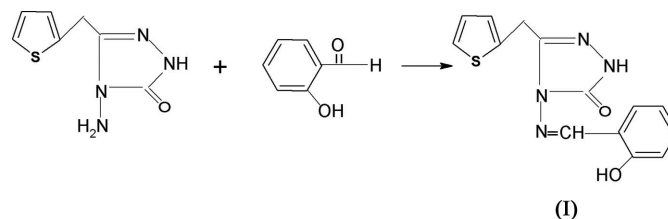
Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.007 \text{ \AA}$
R factor = 0.066
wR factor = 0.185
Data-to-parameter ratio = 14.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{14}\text{H}_{12}\text{N}_4\text{O}_2\text{S}$, the triazole ring is nearly coplanar with the phenol unit, the dihedral angle being $6.60(3)^\circ$. The crystal structure involves an intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond and intramolecular $\text{O}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

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Comment

Triazole ring systems are typical planar six- π -electron partially aromatic systems and 1,2,4-triazole and its derivatives are used as starting materials for the synthesis of many heterocycles (Desenko, 1995). Di- or trisubstituted 1,2,4-triazole derivatives have also been reported to show antitubercular activities (İkizler *et al.*, 1998). In a previous paper, we reported that some 1,2,4-triazol-5-one compounds have antimicrobial effects (Demirbas *et al.*, 2004). The coordination chemistry of azoles acting as ligands for the production of organometallic compounds in the context of modelling biological systems has attracted much interest (İkizler & Sancak, 1992). In this paper, we report the crystal structure of the title compound, (I).



The title compound contains three rings, *viz.* the 1,2,4-triazole ring, *A*, the thiophene ring, *B*, and the phenol ring, *C* (Fig. 1). The dihedral angles between rings *A/B*, *A/C* and *B/C* are $67.39(15)$, $6.60(3)$ and $64.76(1)^\circ$, respectively. These values indicate that the triazole ring is nearly coplanar with the phenol group. The $\text{C}8=\text{O}2$ bond length (Table 1) is comparable with those of similar $\text{C}=\text{O}$ double bonds found in 1,2,4-triazole rings (Arslan *et al.*, 2004; Ocak, Kahveci *et al.*, 2003; Ocak, Çoruh *et al.*, 2003).

In the crystal structure of (I), a strong intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond and intramolecular $\text{O}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds are observed (Table 2 and Fig. 2).

Experimental

A mixture of 4-amino-5-(thien-2-ylmethyl)-2,4-dihydro-1,2,4-triazol-3-one (0.19 g, 0.001 mol) and salicylaldehyde (0.10 ml, 0.12 g, 0.001 mol) was warmed at 413–423 K for 1 h. The solid crude product was recrystallized from ethanol. The crystals were recrystallized several times from the same solvent and were then dried *in vacuo*

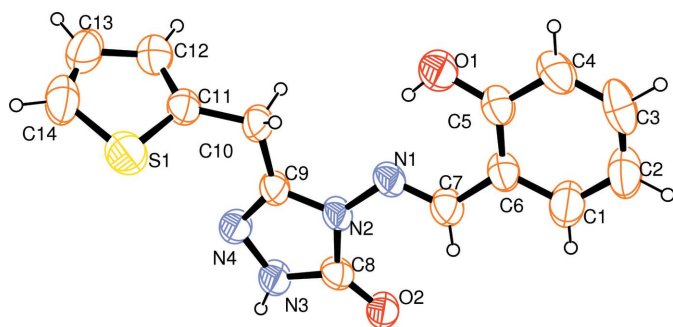


Figure 1

The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

(yield 0.23 g, 80.43%). Spectroscopic analysis: IR (ν , cm^{-1}): 3166 (N–H), 3045 (aromatic C–H), 1711 (C=O), 1618 (C=N), 1606 (C=C); ^1H NMR: δ 4.26 (s, tyf-CH₂), 7.82–7.87 (*m*, aromatic H), 7.34–7.40 (*m*, 3H), 6.91–6.97 (*m*, 4H), 9.95 (*m*, N=CH), 11.95 (*m*, NH), 10.32 (*s*, OH); MS: M^+ 300.92.

Crystal data

$\text{C}_{14}\text{H}_{12}\text{N}_4\text{O}_2\text{S}$
 $M_r = 300.34$
 Triclinic, $P\bar{1}$
 $a = 5.5879$ (7) Å
 $b = 9.2167$ (12) Å
 $c = 14.3000$ (19) Å
 $\alpha = 77.528$ (10)°
 $\beta = 84.181$ (11)°
 $\gamma = 77.290$ (10)°

$V = 700.36$ (16) Å³
 $Z = 2$
 $D_x = 1.424$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.24$ mm⁻¹
 $T = 293$ (2) K
 Plate, yellow
 $0.50 \times 0.21 \times 0.04$ mm

Data collection

Stoe IPDS-2 diffractometer
 ω scans
 Absorption correction: integration
 (*X-RED*; Stoe & Cie, 2002)
 $T_{\min} = 0.932$, $T_{\max} = 0.993$

10648 measured reflections
 2752 independent reflections
 1322 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.176$
 $\theta_{\text{max}} = 26.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.066$
 $wR(F^2) = 0.185$
 $S = 0.87$
 2752 reflections
 190 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0823P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.32$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.24$ e Å⁻³
 Extinction correction: *SHELXL97*
 (Sheldrick, 1997)
 Extinction coefficient: 0.025 (2)

Table 1

Selected geometric parameters (Å, °).

C5–O1	1.347 (5)	C11–S1	1.708 (4)
C8–O2	1.227 (5)	C14–S1	1.692 (6)
C9–N4	1.289 (4)	N1–N2	1.383 (4)
C9–N2	1.371 (5)	N3–N4	1.386 (5)
C9–N2–C8	108.7 (3)	C14–S1–C11	91.6 (3)

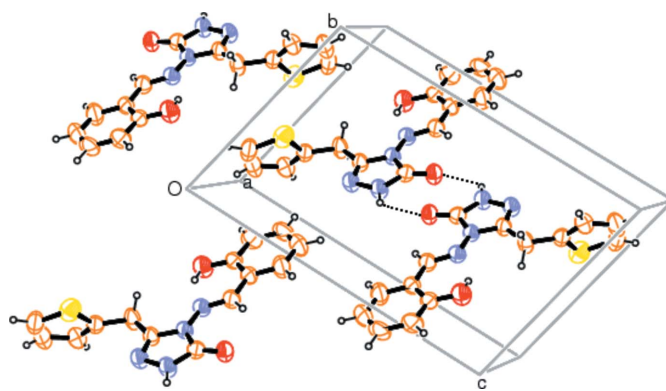


Figure 2

A packing diagram for (I), viewed approximately along the *a* axis. The N–H...O hydrogen bonds are indicated by dashed lines.

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
N3–H7...O2 ⁱ	0.86	1.95	2.791 (4)	167
O1–H5...N1	0.82	1.92	2.639 (4)	146
C7–H6...O2	0.93	2.28	2.946 (5)	128

Symmetry code: (i) $-x, -y + 1, -z + 1$.

The high value of R_{int} indicates that the overall quality of the data may be poor due to the crystal quality. All H atoms were placed in calculated positions, with C–H = 0.93–0.97 Å, N–H = 0.86 Å and O–H = 0.82 Å, and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C,N})$ or $1.5U_{\text{eq}}(\text{O})$.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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